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energy. the energy an object has because of its motion. kinetic theory. a theory explaining the states of matter, based on the concept that all matter consists of tiny particles that are in constant motion. kinetic theory for gases

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## **States of Matter - Atomic Bonding | Interaction Potential**

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element in the same physical state l. the conversion of a liquid to a gas or vapor at a temperature below the boiling point m. states that the tiny particles in all forms of matter are in constant motion n. the change of a solid to a vapor without passing through the liquid state C olumn A amorphous unit cell crystal normal boiling point barometer

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Almost all substances can be classified into three states of matter – solids, liquids and gases. Each state has different properties. Heating and cooling a substance can cause it to change state.

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