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Worksheet Balancing Chemical

Equations - Answer Key Balance the

equations below: 1) $1 \text{ N}_2 + 3 \text{ H}_2 \rightarrow 2 \text{ NH}_3$

2) $2 \text{ KClO}_3 \rightarrow 2 \text{ KCl} + 3 \text{ O}_2$ 3) $2 \text{ NaCl} + 1$

$\text{F}_2 \rightarrow 2 \text{ NaF} + 1 \text{ Cl}_2$ 4) $2 \text{ H}_2 + 1 \text{ O}_2 \rightarrow 2 \text{ H}_2\text{O}$

5) $1 \text{ Pb(OH)}_2 + 2 \text{ HCl} \rightarrow 2 \text{ H}_2\text{O} + 1 \text{ PbCl}_2$

6) $2 \text{ AlBr}_3 + 3 \text{ K}_2\text{SO}_4 \rightarrow 6 \text{ KBr} + 1$

$\text{Al}_2(\text{SO}_4)_3$ 7) $1 \text{ CH}_4 + 2 \text{ O}_2 \rightarrow 1 \text{ CO}_2 + 2$

H_2O 8) $1 \text{ C}_3\text{H}_8 + 5 \text{ O}_2 \rightarrow 3 \text{ CO}_2 + 4 \text{ H}_2\text{O}$

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Balancing Equations Worksheet - 3-13

Worksheet Balancing Word Equations
Chapter 10 (Remember the following are diatomic: H₂, N₂, O₂, F₂, Cl₂, Br₂, I₂)
2) The coefficients should add up to the number at the end that is in parenthesis.
1. carbon + oxygen carbon dioxide (0) 2.

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copper + silver nitrate copper(II) nitrate + silver (4) 3.

Balancing Word Equations Chapter 9 - My Chemistry Class

Balancing Equation And Answer Key - Displaying top 8 worksheets found for this concept.. Some of the worksheets for this concept are Balancing equations

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Balancing Chemical Equations Answer Sheet Worksheets ...

Word Equations Answer Key 1. Zinc and

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Lead (II) nitrate react to form Zinc Nitrate and Lead. $Zn + Pb(NO_3)_2 \rightarrow Zn(NO_3)_2 + Pb$ Single Replacement 2.
Aluminum Bromide and Chlorine gas react to form Aluminum Chloride and Bromine gas. $2AlBr_3 + 3Cl_2 \rightarrow 2AlCl_3 + 3Br_2$ Single Replacement 3.

Word Equations Answer Key Zinc

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and Lead (II) nitrate react ...

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Word Equations Chemistry Worksheet Along with Balancing ...

The first step to balance the equation is to write down the chemical formula of reactants that are listed on the left side of the chemical equation. After this, you can list down the products on the right hand side of the chemical equation.

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There is an arrow between the sides, signaling the direction the reaction is happening in.

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Balancing Chemical Equations Worksheet Answer Key

Balancing Chemical Equations Answer

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Key Balance the equations below: 1) $1 \text{ N}_2 + 3 \text{ H}_2 \rightarrow 2 \text{ NH}_3$ 2) $2 \text{ KClO}_3 \rightarrow 2 \text{ KCl} + 3 \text{ O}_2$ 3) $2 \text{ NaCl} + 1 \text{ F}_2 \rightarrow 2 \text{ NaF} + 1 \text{ Cl}_2$ 4) $2 \text{ H}_2 + 1 \text{ O}_2 \rightarrow 2 \text{ H}_2\text{O}$ 5) $1 \text{ Pb(OH)}_2 + 2 \text{ HCl} \rightarrow 2 \text{ H}_2\text{O} + 1 \text{ PbCl}_2$ 6) $2 \text{ AlBr}_3 + 3 \text{ K}_2\text{SO}_4 \rightarrow 6 \text{ KBr} + 1 \text{ Al}_2(\text{SO}_4)_3$ 7) $1 \text{ CH}_4 + 2 \text{ O}_2 \rightarrow 1 \text{ CO}_2 + 2 \text{ H}_2\text{O}$ 8) $1 \text{ C}_3\text{H}_8 + 5 \text{ O}_2 \rightarrow 3 \text{ CO}_2 + 4 \text{ H}_2\text{O}$ 9) $2 \text{ C}_8\text{H}_{18} + 25 \text{ O}_2 \rightarrow 16 \text{ CO}_2 + 18 \text{ H}_2\text{O}$...

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Chapter 7 Worksheet #1 Balancing Chemical Equations

Balance the following equations: 1. $\text{Al} + \text{N}_2 \rightarrow \text{AlN}$ 2.

BALANCING CHEMICAL EQUATIONS - PRACTICE SHEET # 2

WS6-2 Word Equations.doc 3 Answers to

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Worksheet 6.1 Writing and Balancing Equations 1. Write the chemical equations and balance each of the following word equations. a) Aluminum metal reacts with iron (II) oxide powder to produce aluminum oxide solid and iron metal. $2\text{Al}(s) + 3\text{FeO}(s) \rightarrow \text{Al}_2\text{O}_3(s) + 3\text{Fe}(s)$

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Worksheet 6.2 Word Equations

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