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Chapter 16 Homework Solutions. 17

Aldehydes can be oxidized to carboxylic acids; ketones cannot. The Tollens test and Benedicts test take advantage of this. Both

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tests involve reagents which react with aldehydes to produce a visible product. No reaction occurs with ketones. 18 See the solution in the text book (page A-30).

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reaction has a natural

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tendency to occur and takes place without the continual input of energy from an external source. 3. (a) spontaneous; (b) nonspontaneous; (c) spontaneous; (d) nonspontaneous; (e) spontaneous; (f) spontaneous. 5.

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transfers of energy as heat that accompany ch.... The energy absorbed or released as heat in a chemical or physi.... A measure of the average kinetic energy of the particles in a....

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studying this chapter,
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gain knowledge about
various topics and sub-
topics related to the
chapter such as
postulates of Werner's
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central atom/ ion,
ligand, coordination
number, coordination
...

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1) define: a) buffer b)
buffering capacity c)
buffering range d)
indicator e)
equivalence point f)
solubility g) molar
solubility 2) A buffer is

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KClO and 0.330 M
HClO. With HClO
having a $K_a = 2.9 \times 10^{-8}$. What is the pH of
the solution?

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