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6. a) NH₄⁺ + b) OH⁻ - c) A conjugate acid-base pair are two chemical species that differ by a proton. ChemActivity 42 Answers to Problems and Exercises 1. SO₄²⁻; CO₃²⁻; OH⁻; O²⁻; H₂O; NH₃; CH₃NH₂; F⁻; CH₃COO⁻; 2. HSO₄⁻; HCO₃⁻; H₃O⁺; H₂O; OH⁻; NH₄⁺; CH₃NH₃⁺; HCN; CH₃COOH; HF; H₂CO₃; NH₃.

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4.The bond energy (in MJ/mole) is the energy required to break one mole of the specified bonds. Triple bonds share six electrons between two atoms and are stronger (have greater bond energy) than double bonds, which share four electrons between two atoms.

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