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Chemistry in Context -
Table of Contents.

Menu. Interactives. Ch.
1 | Portable Electronics:
The Periodic Table in
the Palm of Your Hand.

Ch. 2 | The Air We
Breathe. Ch. 3 |
Radiation from the
Sun. Ch. 4 | Climate
Change. Ch. 5 | Water
Everywhere. Ch. 6 |
Energy from
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- Table of Contents -
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Water for Life 40
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Hydrated means
chemically bonded to
water or containing
water. b. Anhydrous
means containing no
water or without water.

c. Water of

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crystallisation is the water present in the crystals of certain compounds such as Epsom salts. 2 The crystals of Epsom salts are almost colourless.

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someone has tried to

trick you with this

question: "Which is

heavier, a pound of

lead or a pound of

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feathers?" Many people would instinctively answer "lead." When they give this incorrect answer, these people are really thinking of density. If a piece of lead and a feather of the same volume are weighed, the lead ...

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...

Doing chemistry
Page 14/24

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involves attempting to answer questions and explain observations in terms of the laws and theories of chemistry, using procedures that are accepted by the scientific community. There is no single route to answering a question or explaining an observation, but there is an aspect common to every approach: Each uses knowledge based on experiments that can

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be reproduced to verify the results.

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1. The protons in the nucleus do not change during normal chemical reactions. Only the outer electrons move. Positive charges form when electrons are lost. 3. P, I, Cl, and O would form anions because they are nonmetals. Mg, In, Cs,

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Pb, and Co would form cations because they are metals. 5.

**Answer Key Chapter
7 - Chemistry 2e |
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Definitions for the book
Chemistry in Context;
Chapter 5; UCF CHM
1020. Terms in this set
(41) Electronegativity.
is a measure of the
attraction of an atom
for an electron in a
chemical bond. Polar
Covalent Bond. a

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covalent bond in which the electrons are not equally shared but rather are closer to the more electronegative atom.

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chemistry in the
context of their own
lives and significant
issues facing science
and the world.

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Doing chemistry
involves attempting to
answer questions and
explain observations in
terms of the laws and

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theories of chemistry, using procedures that are accepted by the scientific community. There is no single route to answering a question or explaining an observation, but there is an aspect common to every approach: Each uses knowledge based on experiments that can be reproduced to verify the results.

1.1 Chemistry in *Page 20/24*

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Context | General College Chemistry I

The non-traditional approach of Chemistry in Context reflects today's technological issues and the chemistry principles within them. Global warming, alternate fuels, nutrition, and genetic engineering are examples of issues that are covered in Chemistry in Context .

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