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4. Wind D. When two objects are at the same temperature. 4. Solid D. Allows convection; can be a good conductor of heat. 5. Liquid E. Any material that resists the movement of heat through it. 6. Gas F. No convection can occur in this. 2. Thermal Equilibrium 5. Convection 6. Thermodynamics E. The study of how heat moves. F.

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HW Unit 8:6 — Thermodynamics Mr. Murray, IPC

cstephenmurray.com 1. ConDuction, ConVection, or Radiation?

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A. ___ You pick up a hot piece of metal and get burned. B. ___ You put your hand above a pan of hot water. C. ___ You feel the heat from a brick wall when you put your hand next to the wall, but not touching it. D.

HW Unit 8:6 — Thermodynamics A-day: Due Mon., 3/5 (Assig ...

Thermodynamics Thermodynamics is the study of how heat moves. Heat always transfers from hot to cold. Heat does not rise (hot air rises). 25 oC 5 oC Heat transfer Conduction Conduction transfers heat through objects touching. Insulators slow down heat transfer. Materials with air pockets are good insulators. 25oC 5oC slow transfer wood

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Unit 8: Temperature, Heat, & Thermodynamics

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1+ ammonium NH_4^+ 1- acetate $\text{C}_2\text{H}_3\text{O}_2^-$ 2- carbonate CO_3^{2-}
2-2- chromate CrO_4^{2-} 2-1- hydrogen carbonate HCO_3^- 1-1+ hydronium H_3O^+ 1- hydroxide OH^- 1-1- nitrate NO_3^- 1-2- peroxide O_2^{2-} 2-3- phosphate PO_4^{3-} 2-2- sulfite SO_3^{2-} 2- sulfate SO_4^{2-} 2-2- sulfite SO_3^{2-} 2-Greek Prefixes Mono - 1 Di - 2 Tri - 3 Tetra - 4 Penta - 5 Hexa - 6 Hepta - 7 Octa - 8 Nona - 9

Naming Compounds - Mrs. McCutchen's Class

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- The second law of thermodynamics introduces the notion of entropy (S), a measure of system disorder (messiness) • U is the quantity of a system's energy, S is the quality of a system's energy. • Another C.P. Snow expression: - not knowing the 2 nd law of thermodynamics is the cultural equivalent to never having read Shakespeare

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