

# Read Online Modern Chemistry Chapter 5 The Periodic Law Test Answer Key

## Modern Chemistry Chapter 5 The Periodic Law Test Answer Key

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### Modern Chemistry Chapter 5 The

Highlights 5.1 - The periodic law states that the physical and chemical properties of the elements are periodic functions of their atomic numbers. - The periodic table is an arrangements of the elements in order of their atomic numbers so that elements with similar properties fall in the same column.

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the elements are periodic functions of their atomic numbers. periodic table. an arrangement of the elements in order of their atomic numbers so that elements with similar properties fall in the same column, or group. lanthanides.

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### **Modern Chemistry Chapter 5 Test The Periodic Law Answers ...**

Modern Chemistry Chapter 5 - Chemistry Chapter 5 the Periodic ... Unformatted text preview: Chemistry Chapter 5 the Periodic Table In 1860, Stanislao Cannizzaro presented a method for accurately measuring the relative atomic masses of the elements. Chemistry Chapter 5 The Periodic Law In 1869, Dmitri Mendeleev published his first "periodic table".

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## **Modern Chemistry Chapter 5 Review The Periodic Law Answer Key**

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Modern Chemistry 5 The Periodic Law CHAPTER 5 REVIEW The Periodic Law SECTION 2 SHORT ANSWER Use this periodic table to answer the following questions in the space provided. 1. Identify the element and write the noble-gas notation for each of the following: a. the Group 14 element in Period 4

## **Modern Chemistry Chapter 5 The Periodic Law Section 1 Answers**

CHAPTER 5 REVIEW The Periodic Law MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Consider the neutral atom with 53 protons and 74 neutrons to answer the following questions. 53 a. What is its atomic number? 127 b. What is its mass number? atomic number c. Is the element's position in a modern periodic table determined by

## **5 The Periodic Law**

CHAPTER 5 REVIEW The Periodic Law SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design.

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## Modern Chemistry Chapter 5 Review Answers Section 1

CHAPTER 5 REVIEW The Periodic Law SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design.

## Modern Chemistry Chapter 5 Section 3 Review Answers

a. 2 and 4 c. 4 and 5 b. 3 and 4 d. 5 and 6 \_\_\_\_ 21. The discovery of the noble gases changed Mendeleev's periodic table by adding a new. a. period. c. group. b. series. d. sublevel block. \_\_\_\_ 22. In the modern periodic table, elements are ordered according to. a. decreasing atomic mass. c. increasing atomic number. b. Mendeleev's original ...

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one electron from an atom is called its \_\_\_\_\_. 18.

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