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Objectives. After completing this section, you should be able to describe how differences in electronegativity give rise to bond polarity. arrange a given series of the elements most often encountered in organic chemistry (C, H, O, N, S, P and the halogens) in order of increasing or decreasing electronegativity, without referring to a table of electronegativities.

2.1: Polar Covalent Bonds - Electronegativity - Chemistry

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