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Subatomic Particle Gizmo

Put the correct answer in the space provided (you will need a Periodic Table): (3 pts each) 1. What is the atomic number of oxygen (O)? ____ 2. How many protons does an aluminum (Al) atom have? ____ 3. How many electrons does a calcium (Ca) atom have? ____ For #4 - #6, consider a potassium (K) atom with a mass number of 41: 4. How many ...

Subatomic Particles Worksheet

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Subatomic Particle Gizmo Answers Key

Gizmo Warm-up. Atoms are tiny particles of matter that are made up of three particles: protons, neutrons, and electrons. The Element Builder Gizmo™ shows an atom with a single proton. The proton is located in the center of the atom, called the nucleus. 1. Use the arrow buttons () to add protons, neutrons, and electrons to the atom. Press ...

Student Exploration: Element Builder

After completing this I will have students volunteer to answer questions 4-6 on the board. Write an equation for the alpha decay of polonium, and then use the Gizmo to check your answer. What isotope remains after the alpha decay of polonium-212? Write an equation for the alpha decay of radium, and then use the Gizmo to check your answer.

Ninth grade Lesson Day 1: Radioactive Decay Using A Gizmo.

Ch 4 & 25 The Atom and Nuclear Chemistry key How Atoms Differ Parts of Atoms: 1. Most people already know that the atom is made up of three main parts, the protons and neutrons in the nucleus and the electrons somewhere outside the nucleus. 2. Why is an atom electrically neutral? The atom has the same number of protons and electrons #2 Electron

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Name: Catherine Kopf Date: February 15, 2017 Student Exploration: Nuclear Decay NCVPS Chemistry Fall 2014 Vocabulary: alpha particle, atomic number, beta particle, daughter product, gamma ray, isotope, mass number, nuclear decay, positron, radioactive, subatomic particle Prior Knowledge Questions (Do these BEFORE using the Gizmo.) The chart below gives the locations, charges, and approximate ...

M2L4P2NuclearDecayGizmo_CatherineKopf - Name Catherine ...

1. The smallest particle OF an element that retains the properties OF that element 2. A positively charged subatomic particle 3. A negatively charged subatomic particle 4. A particle with no charge 5. The central part OF an atom containing protons and neutrons Match each item with the correct statement: I.

Part A: The Atom & Subatomic Particles

Name: ____ Date: ____ Student Exploration: Nuclear Decay Vocabulary: alpha particle, atomic number, beta particle, daughter product, gamma ray, isotope, mass number, nuclear decay, positron, radioactive, subatomic particle Prior Knowledge Questions (Do these BEFORE using the Gizmo.) The chart below gives the locations, charges, and approximate masses of three subatomic particles.

Nuclear Decay Gizmo - Name Date Student Exploration ...

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Student Exploration Sheet Food Chain Answer Key

The field of subatomic particles has expanded dramatically with the construction of powerful particle accelerators to study high-energy collisions of electrons, protons, and other particles with matter. As particles collide at high energy, the collision energy becomes available for the creation of subatomic particles such as mesons and hyperons.

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